

INTRODUCTION TO REFRIGERATION

- Cooling to preserve products and provide comfort
- 1900s were the beginnings of mechanical refrigeration systems
- Refrigeration process temperature ranges
 - High temperature – Air conditioning (comfort)
 - Medium temperature – Fresh food preservation
 - Low temperature – Frozen food preservation

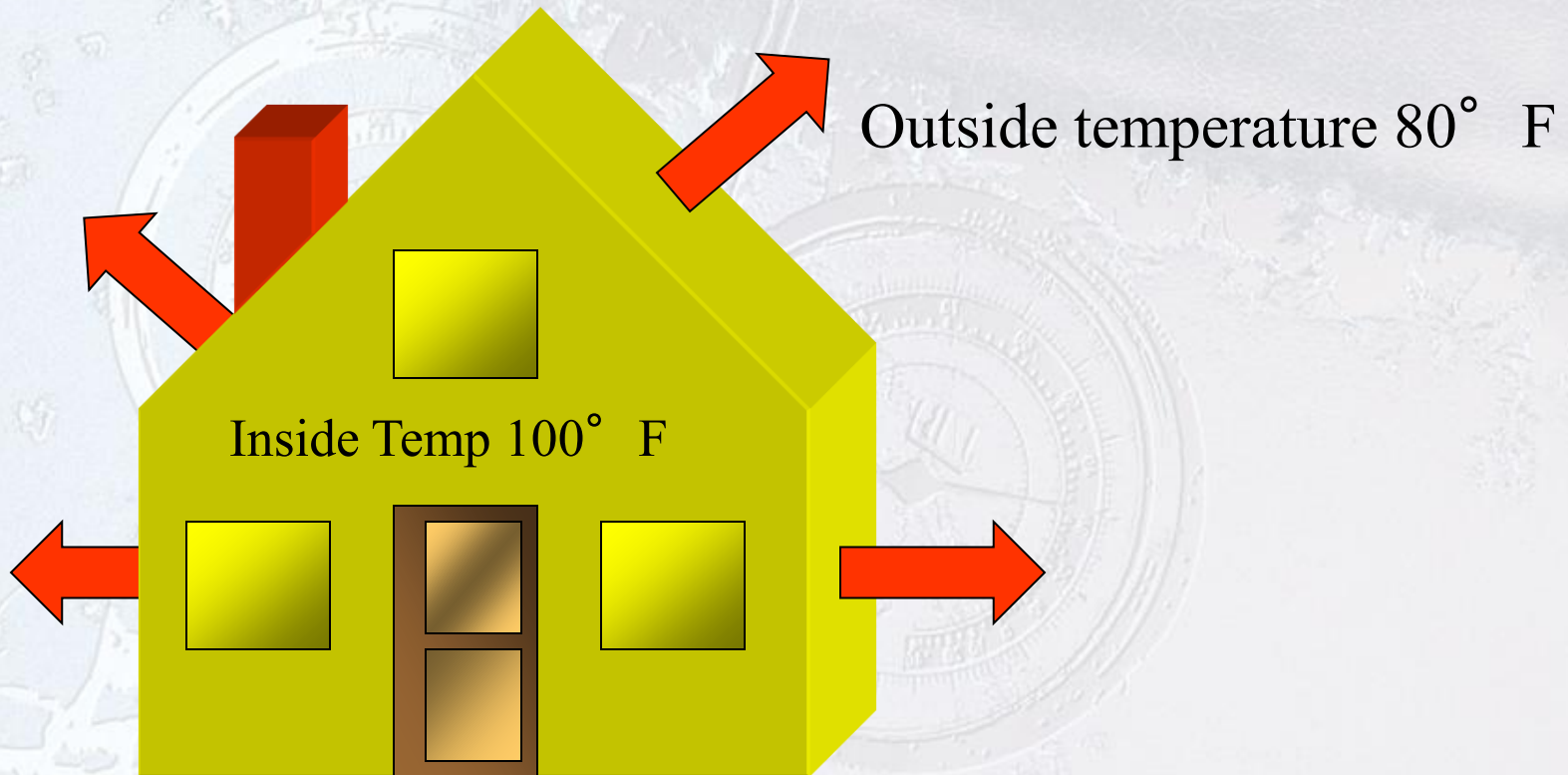
REFRIGERATION

- Process of transferring heat from a place where it is objectionable to a place where it makes little or no difference
- Heat naturally flows from a warmer substance to a cooler substance
- Heat will flow naturally from a 100° F house if the outside temperature is 80° F
- Mechanical refrigeration is needed if the house is 80° F and the outside temperature is 100° F

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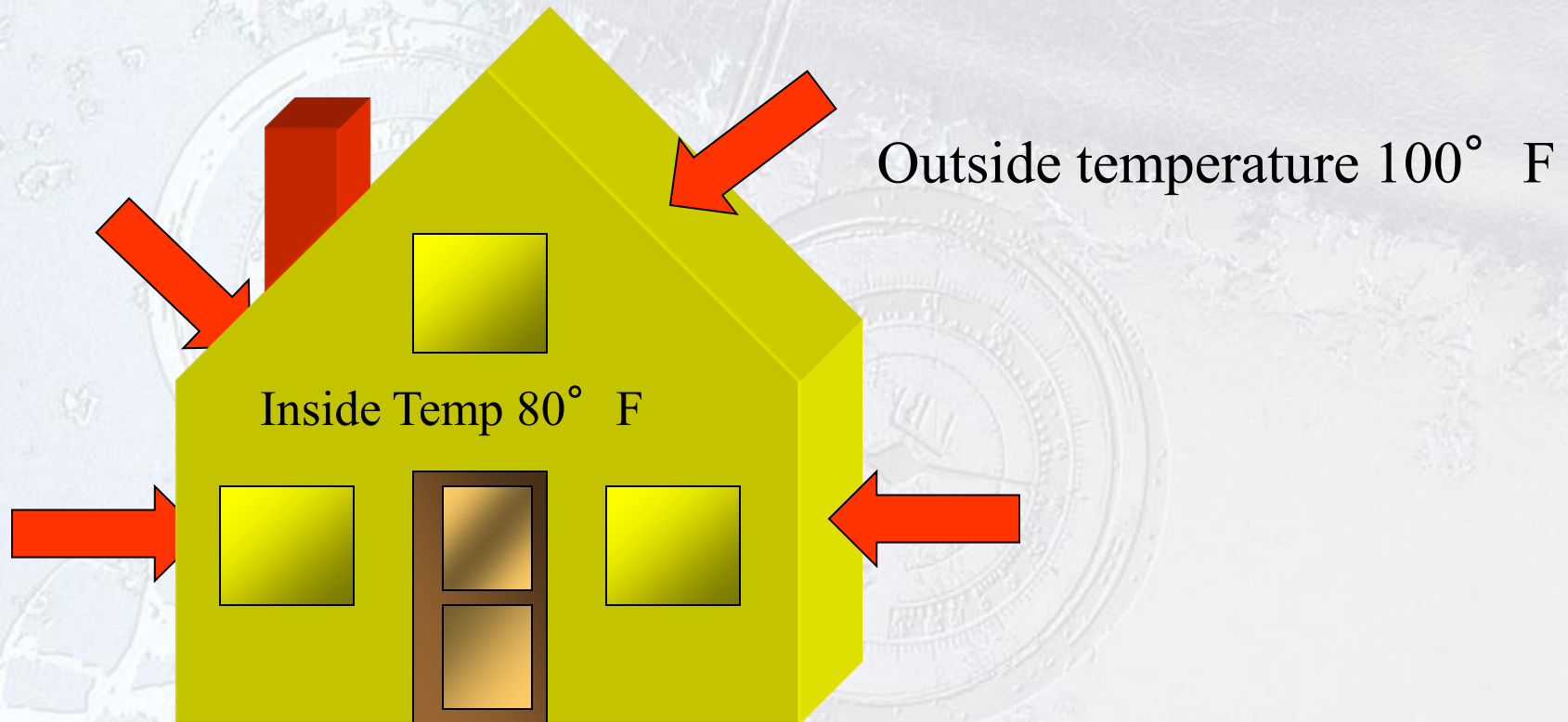
Heat flows naturally from a warmer substance to a cooler substance



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Heat flows naturally from a warmer substance to a cooler substance



Mechanical refrigeration would be needed to cool this house

RATING REFRIGERATION EQUIPMENT

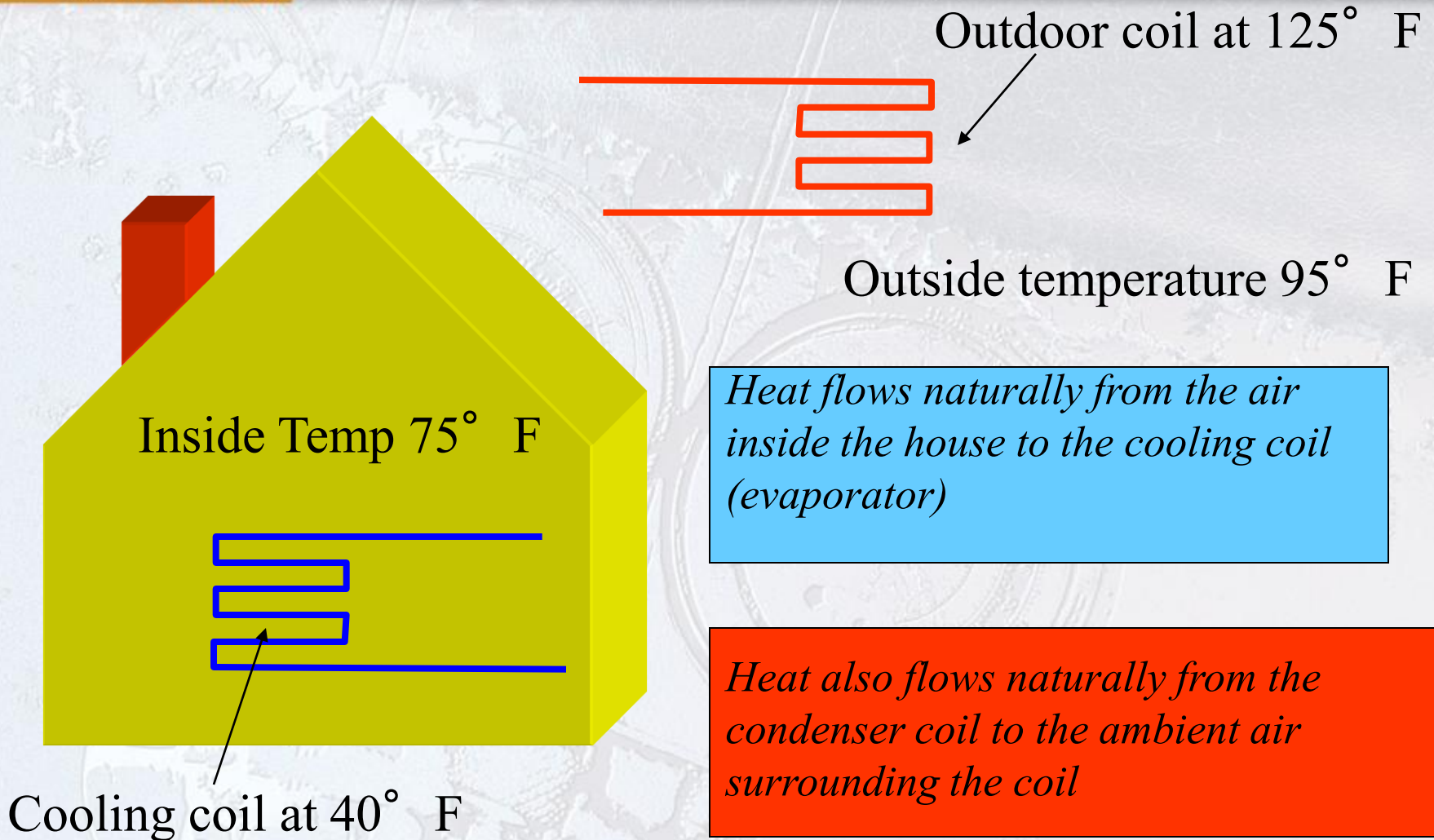
- It takes 144 btu to melt one pound of ice at 32° F
- 2,000 pounds of ice (1 ton) will require 288,000 btu to melt (144 btu x 2,000 pounds)
- If the melting of 1 ton of ice takes place in one day (24 hours), 12,000 btu must be absorbed by the ice every hour (288,000 btu / 24 hours)
- 12,000 btu/hr = 200 btu/min = 1 ton of refrigeration

THE REFRIGERATION PROCESS

- Heat is pumped from a cool box to a warm room
- Pumping of heat is similar to pumping water uphill
- Air conditioners pump heat from inside to the outside
 - Inside temperature 75° F, Outside temperature 95° F
 - Cooling (indoor) coil temperature 40° F
 - Condenser (outdoor) coil temperature 125° F
 - Indoor heat travels to the indoor coil
 - System heat flows from the outdoor coil to the outside air

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TEMPERATURE AND PRESSURE RELATIONSHIP

- Water boils at 212° F at atmospheric pressure (29.92 in. Hg)
- Water boils at 250° F if pressure is increased to 15 psig
- Water boils at 40° F if pressure is reduced to 0.248 in. Hg
- Refrigerants are substances that boil at low pressures and temperatures and condense at high pressures/temperatures
- Saturation temperature – Point at which the addition or removal of heat will result in a change of state
- During a change of state, the temperature remains constant

REFRIGERATION COMPONENTS

- Evaporator - Absorbs heat from area to be cooled
- Compressor – Creates pressure difference needed to facilitate refrigerant flow through the system
- Condenser - Rejects system heat
- Metering device – Regulates refrigerant flow to the evaporator

THE EVAPORATOR

- Heat exchange surface used to absorb heat
- Located on the low-pressure side of the system between the metering device and the compressor
- Operates at temperatures lower than the medium being cooled or conditioned
- Absorbs heat by boiling a low temperature liquid into a low temperature vapor

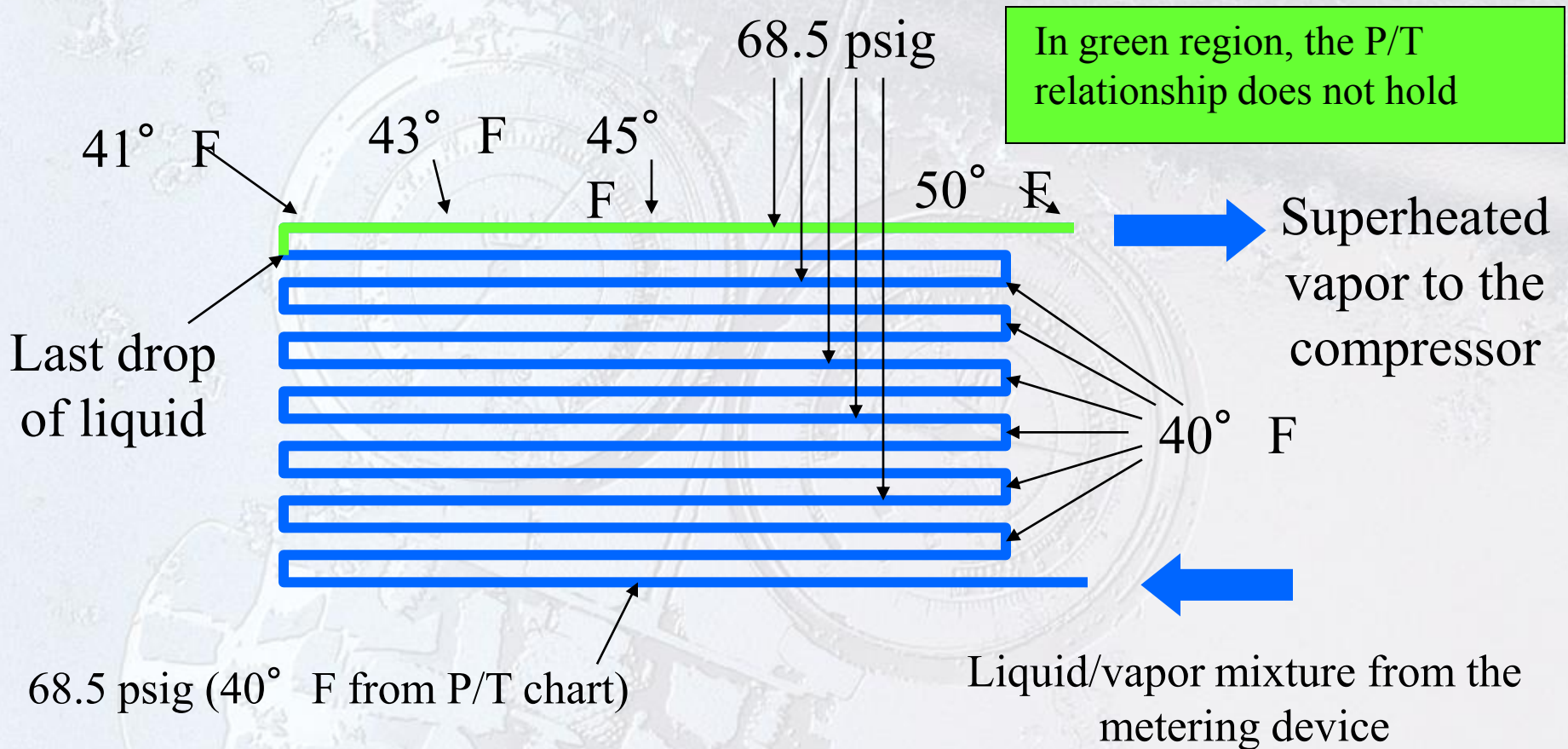
THE EVAPORATOR (cont'd)

- Refrigerant typically enters the evaporator as a liquid/vapor mix (75% liquid, 25% vapor)
- Superheat
 - The heating of a vapor above its saturation temperature
 - Ensures that no liquid refrigerant enters the compressor
 - Equal to the evaporator outlet temperature minus the evaporator saturation temperature
 - Design superheat is typically between 8° F and 12° F
- Superheated vapor does not follow a pressure/temperature relationship

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AIR CONDITIONING APPLICATION: R-22 EVAPORATOR



THE COMPRESSOR

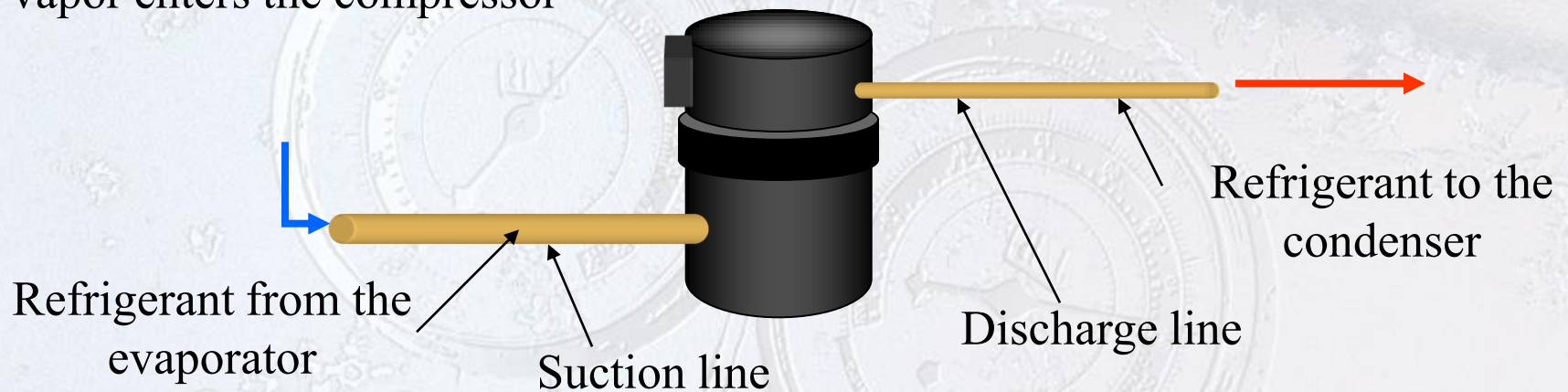
- Pumps heat-laden vapor from the evaporator to the condenser by increasing the refrigerant pressure
- Reduces pressure on the low-side of the system
- Increases pressure on the high-side of the system
- Common compressor types include the scroll, reciprocating and the rotary
- Positive displacement compressors require that the compressed gas be moved to the condenser

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Low pressure, low temperature superheated vapor enters the compressor

High pressure, high temperature superheated vapor leaves the compressor



Refrigerant in the compressor is superheated, so it does not follow a pressure/temperature relationship!

THE CONDENSER

- Rejects sensible and latent heat from the system that was absorbed by compressor and evaporator
- Located on the high-pressure side of the system
- The refrigerant condenses from a high temperature vapor to a high temperature liquid
- Condensing temperature is determined by the high side pressure in the system
- Refrigerant is subcooled at the outlet of the condenser

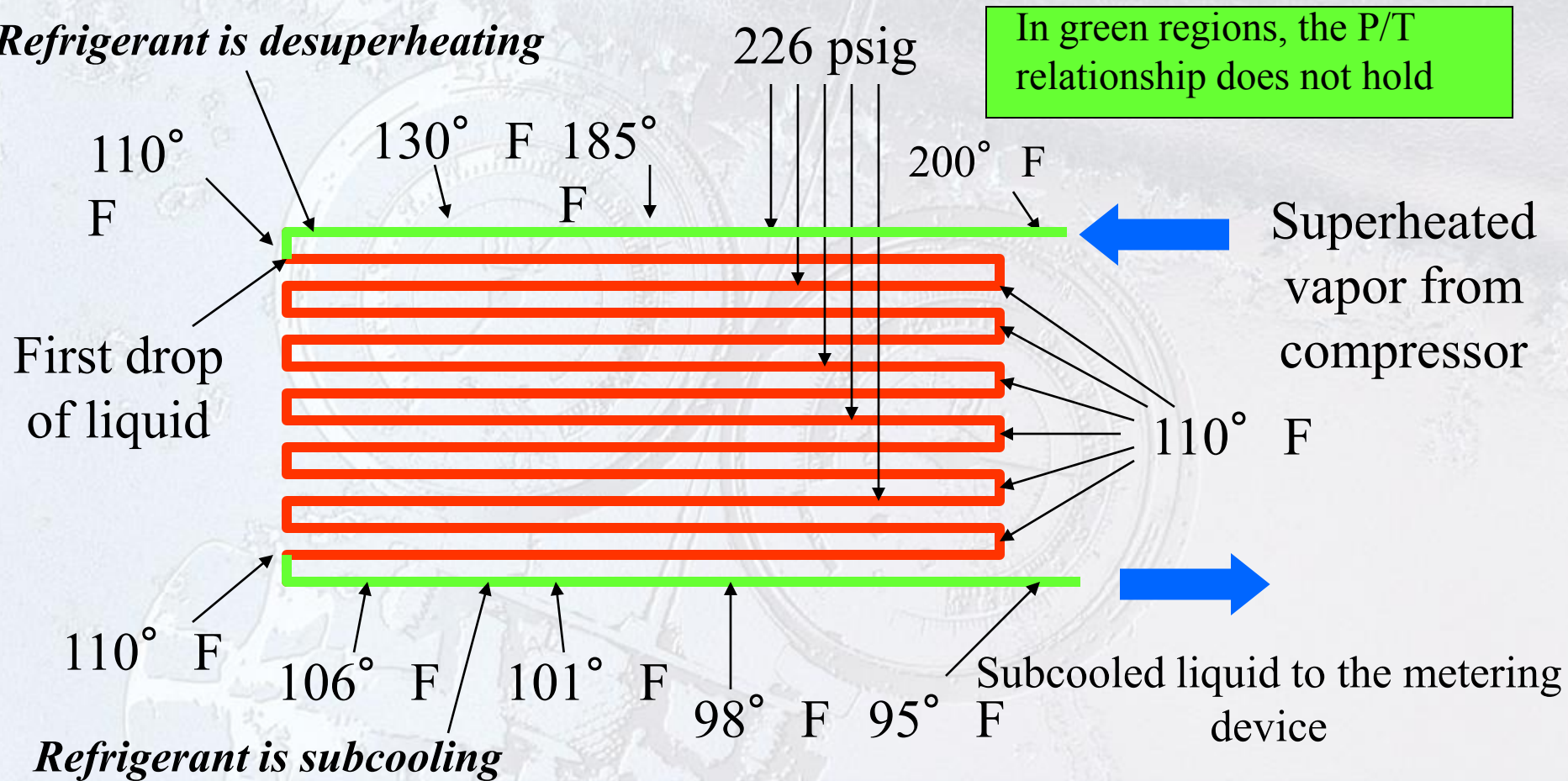
THE CONDENSER (cont'd)

- Subcooling
 - The cooling of liquid refrigerant below its saturation temperature
 - Standard air-cooled systems are designed to operate with a minimum of 10° F of subcooling
 - High efficiency condensers operate with more subcooling than standard efficiency systems
 - Determined by subtracting the condenser saturation temperature from the condenser outlet temperature

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AIR CONDITIONING APPLICATION: R-22 CONDENSER



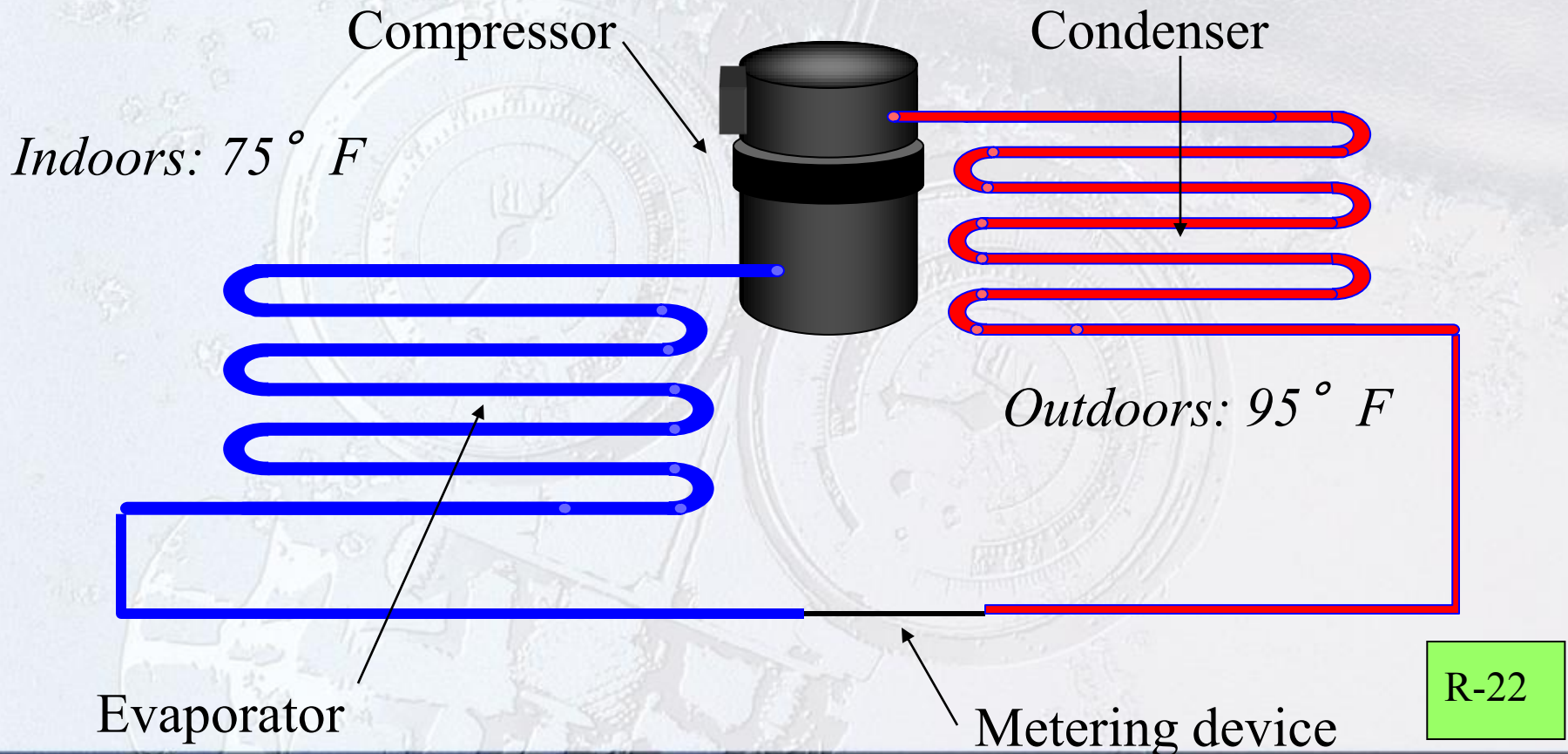
THE METERING DEVICE

- Controls the flow of subcooled liquid from the condenser to the evaporator
- Creates a pressure drop between the high and low pressure sides of the system
- About 25% of the liquid leaving the metering device immediately vaporizes (flash gas)
- Three commonly used metering devices are the capillary tube, automatic expansion valve and the thermostatic expansion valve

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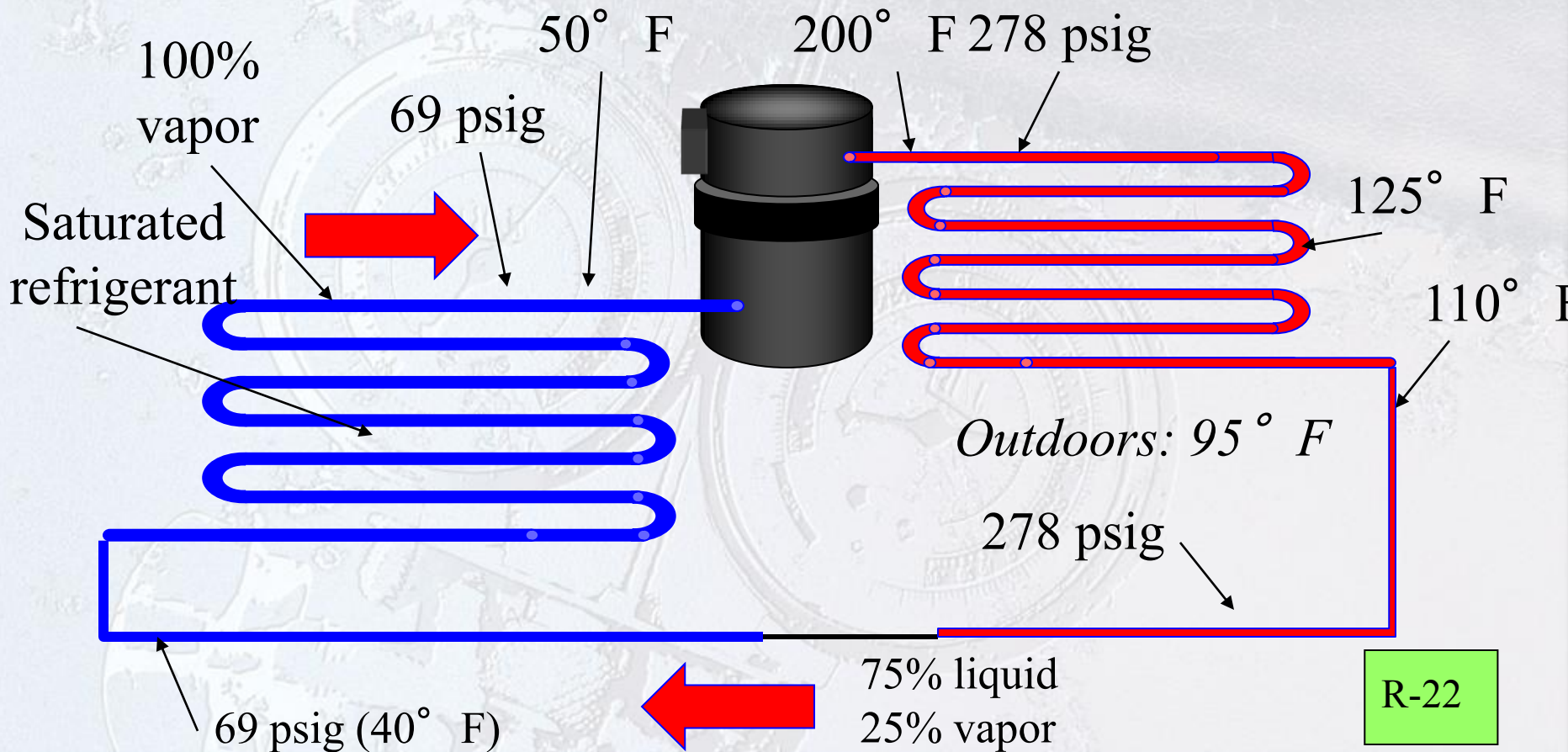
PUTTING IT TOGETHER: A WINDOW UNIT EXAMPLE



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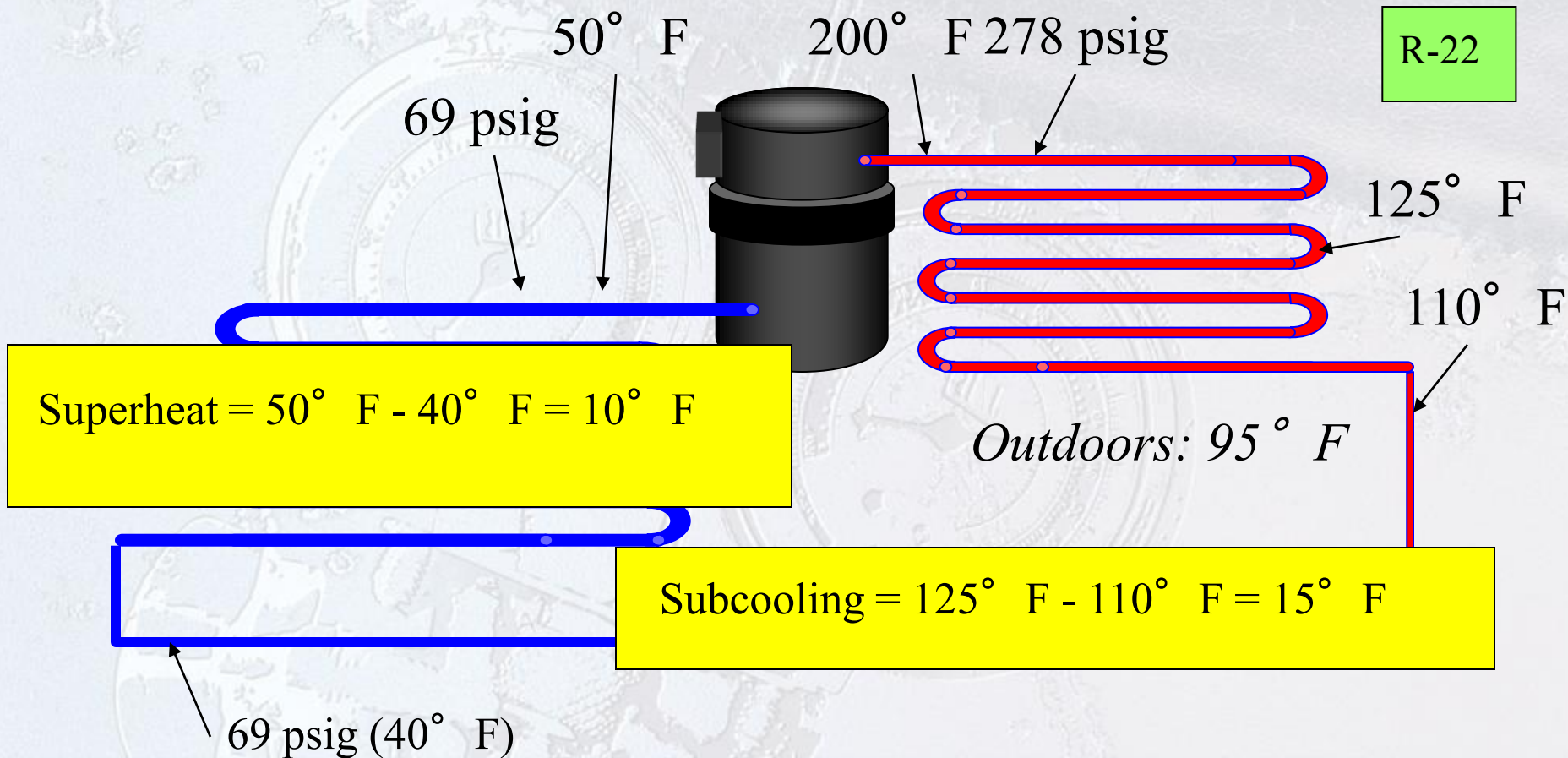
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PUTTING IT TOGETHER: A WINDOW UNIT EXAMPLE



REFRIGERANTS

- R-12 used primarily for high and medium temperature refrigeration applications (manufacture banned in 1996)
- R-22 used primarily in air conditioning applications (slated for total phase-out in 2030)
- R-500 and R-502 banned in 1996
- R-134a – replacement for R-12 with retrofit
- Replacements for R-22 include R-410a and R-407c

REFRIGERANTS MUST BE SAFE

- Designed to protect people from sickness, injury and death
- Proper ventilation is required
- Refrigerants can displace oxygen if permitted to accumulate
- Modern refrigerants are non-toxic
- When burned, toxic/corrosive gases are created

REFRIGERANTS MUST BE DETECTABLE

Methods used for detecting refrigerant leaks include:

- Soap bubble solution – pinpoints leaks
- Halide leak detector – uses an open flame
- Electronic leak detectors – general area leaks
- Ultraviolet leak detectors – pinpoints leaks
- Ultrasonic leak detectors – uses sound waves

GENERAL REFRIGERANT NOTES

- Refrigerants should boil at low temperatures at atmospheric pressure so that low temperatures can be obtained without going into a vacuum
- It is illegal to intentionally vent refrigerant to the atmosphere (Stiff fines for violations)
- Mandatory certification for technicians
- The EPA set refrigerant phase-out schedules
- Refrigerant cylinders and drums are color-coded

RECOVERY, RECYCLING AND RECLAIMING OF REFRIGERANTS

- Refrigerant recovery is mandatory during service and installation operations
- Intended to reduce the emissions of CFC, HCFC and HFC refrigerants
- Recovery equipment must be used according to manufacturer's instructions

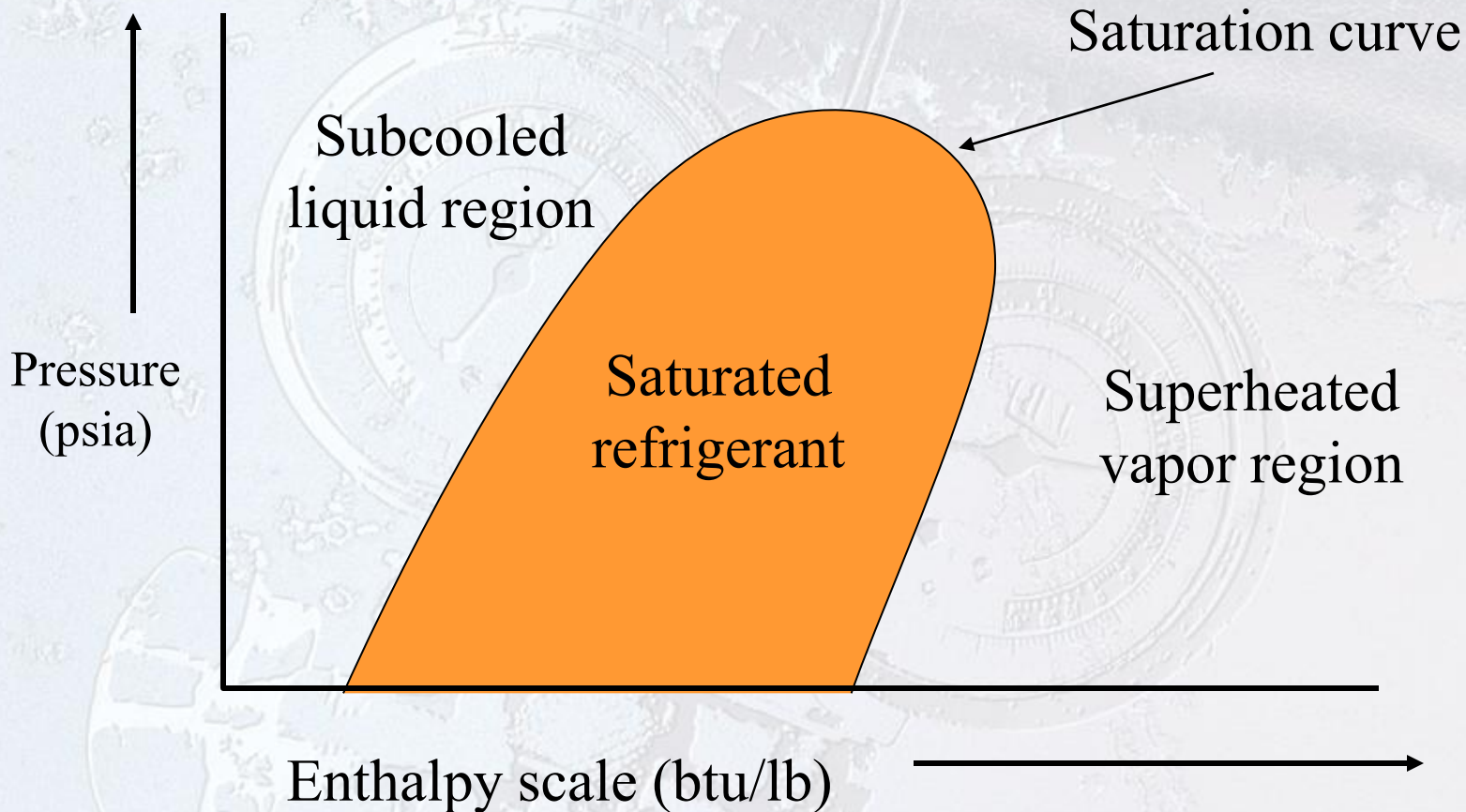
PLOTTING THE REFRIGERATION CYCLE

- Pressure-enthalpy chart
 - Used to create a graphical representation of a system
 - Pressure scales on the vertical axis (psia)
 - Enthalpy scale along the bottom of the chart
 - Horseshoe curve represents the saturation curve
 - Refrigerant is saturated on and under the curve
- Enthalpy is defined as heat content

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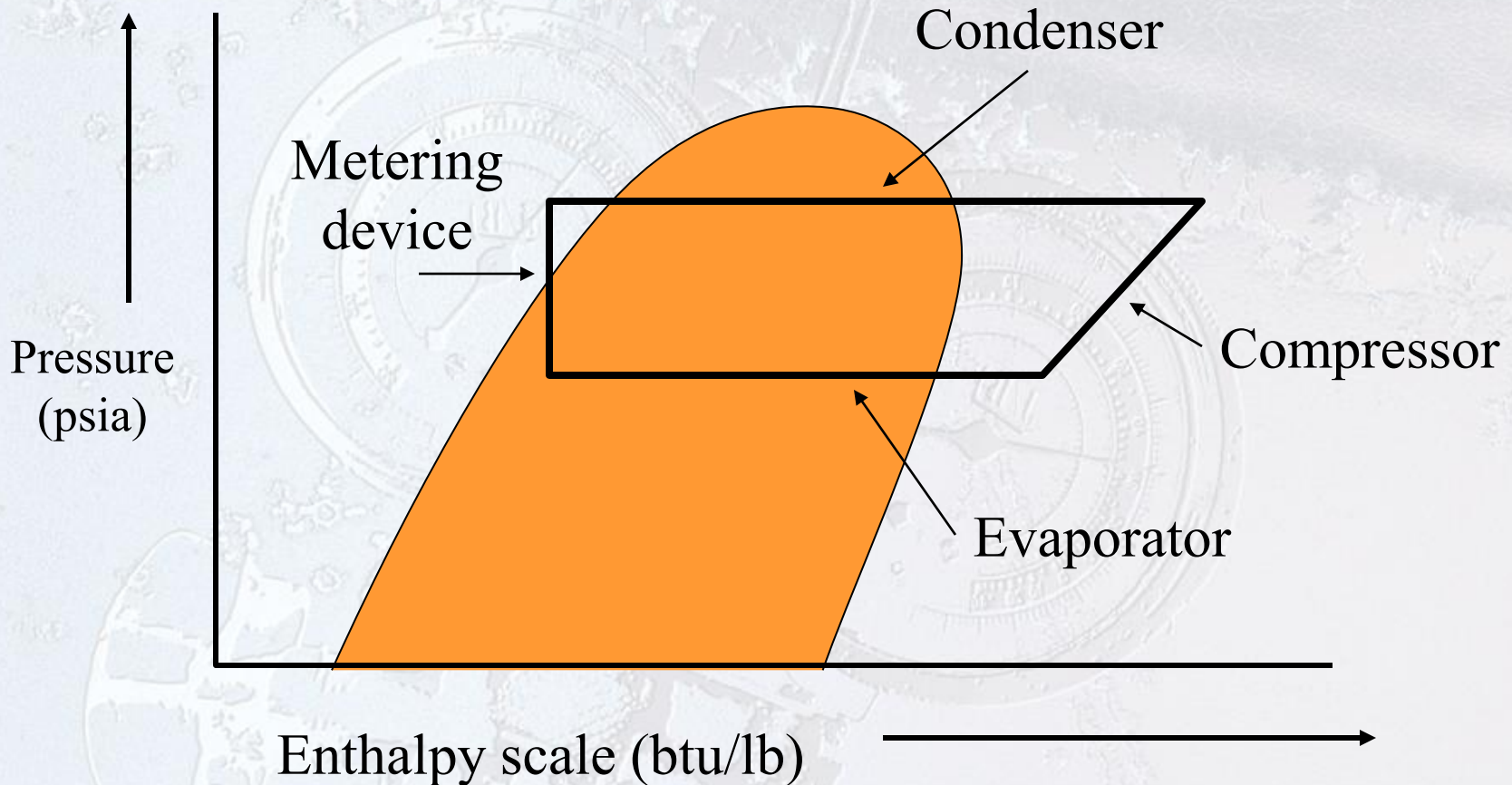
THE PRESSURE ENTHALPY CHART



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THE PRESSURE ENTHALPY CHART: A SAMPLE PLOT



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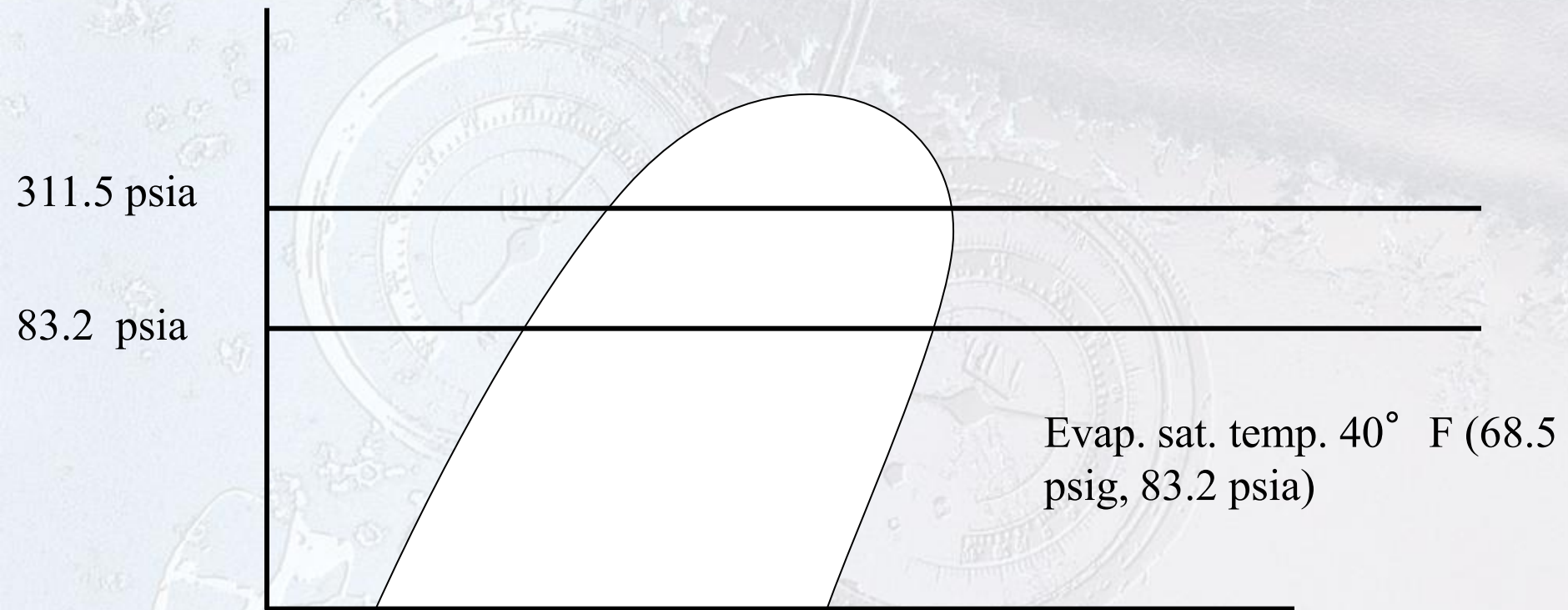
THE PRESSURE ENTHALPY CHART: AN R-22 EXAMPLE



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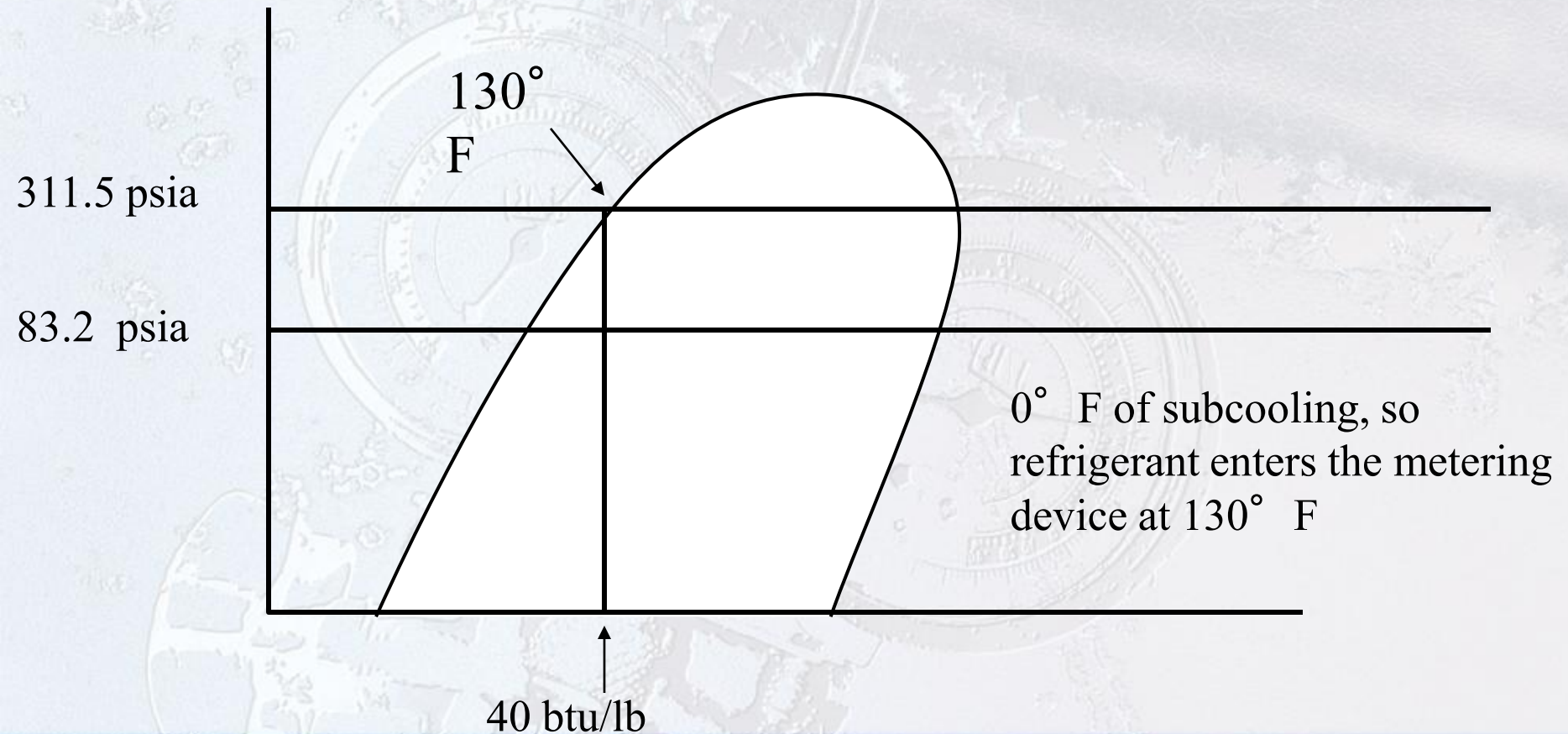
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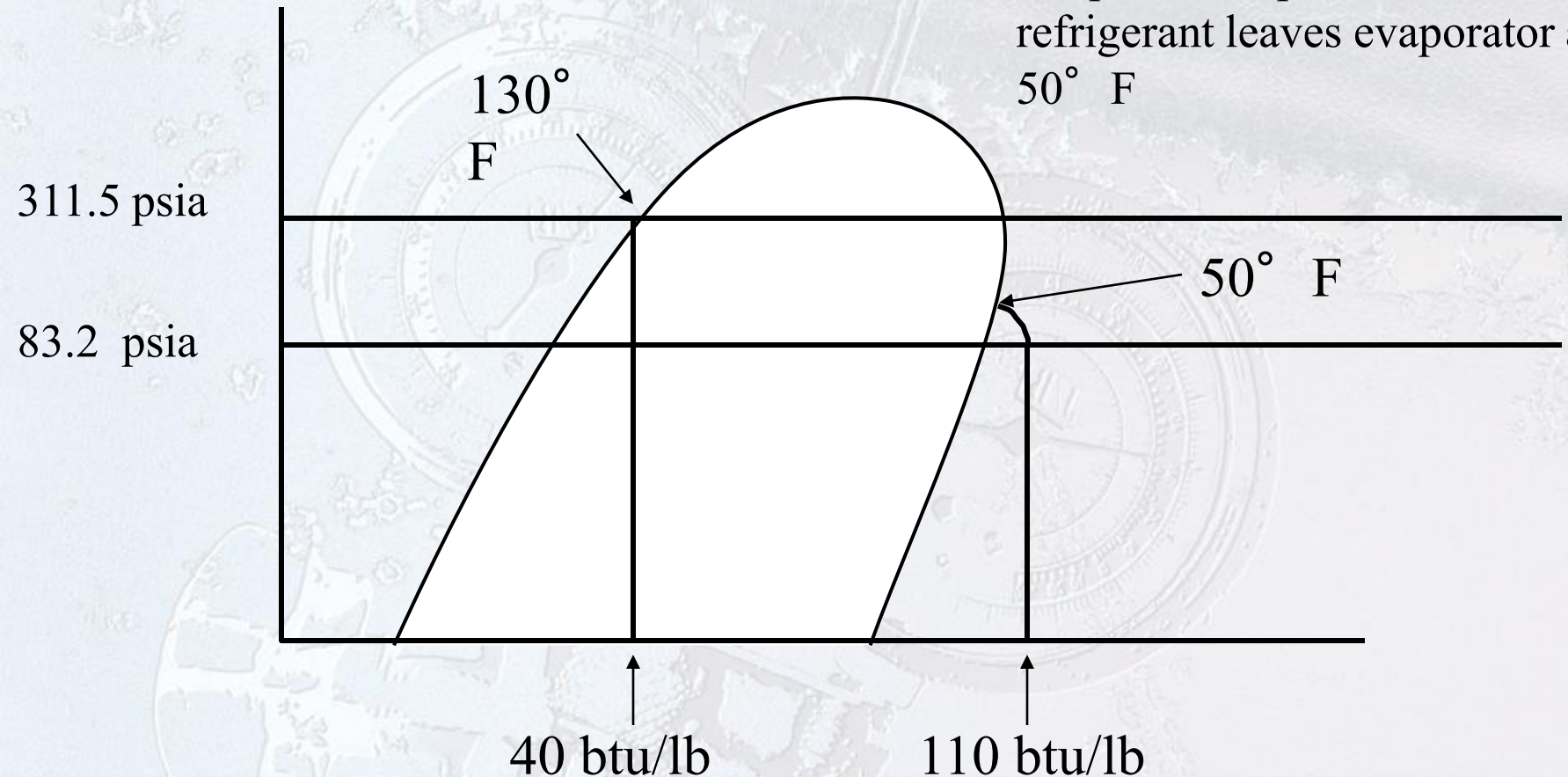


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THE PRESSURE ENTHALPY CHART: AN R-22 EXAMPLE

Evaporator superheat is 10° F, so refrigerant leaves evaporator at 50° F

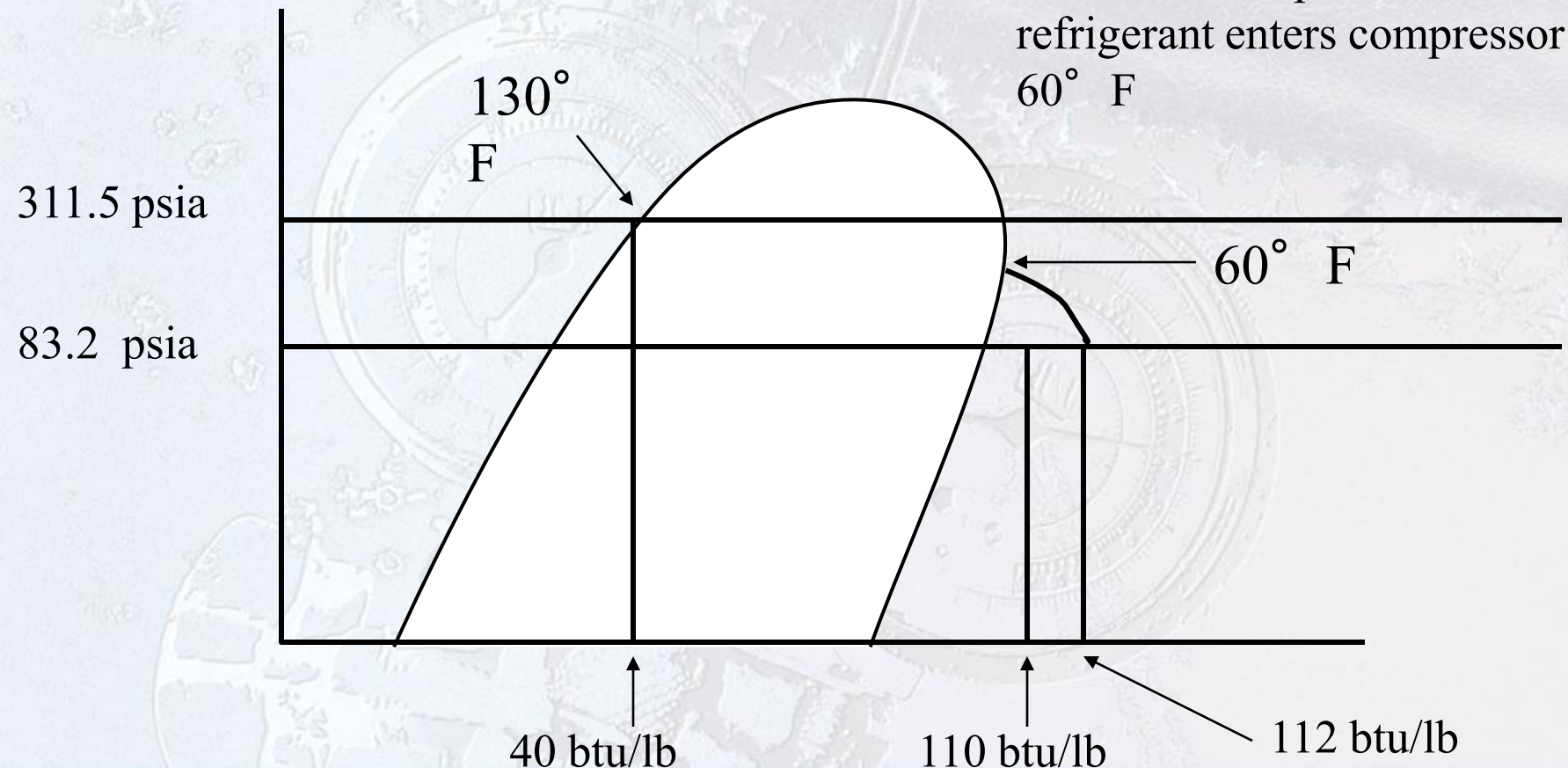


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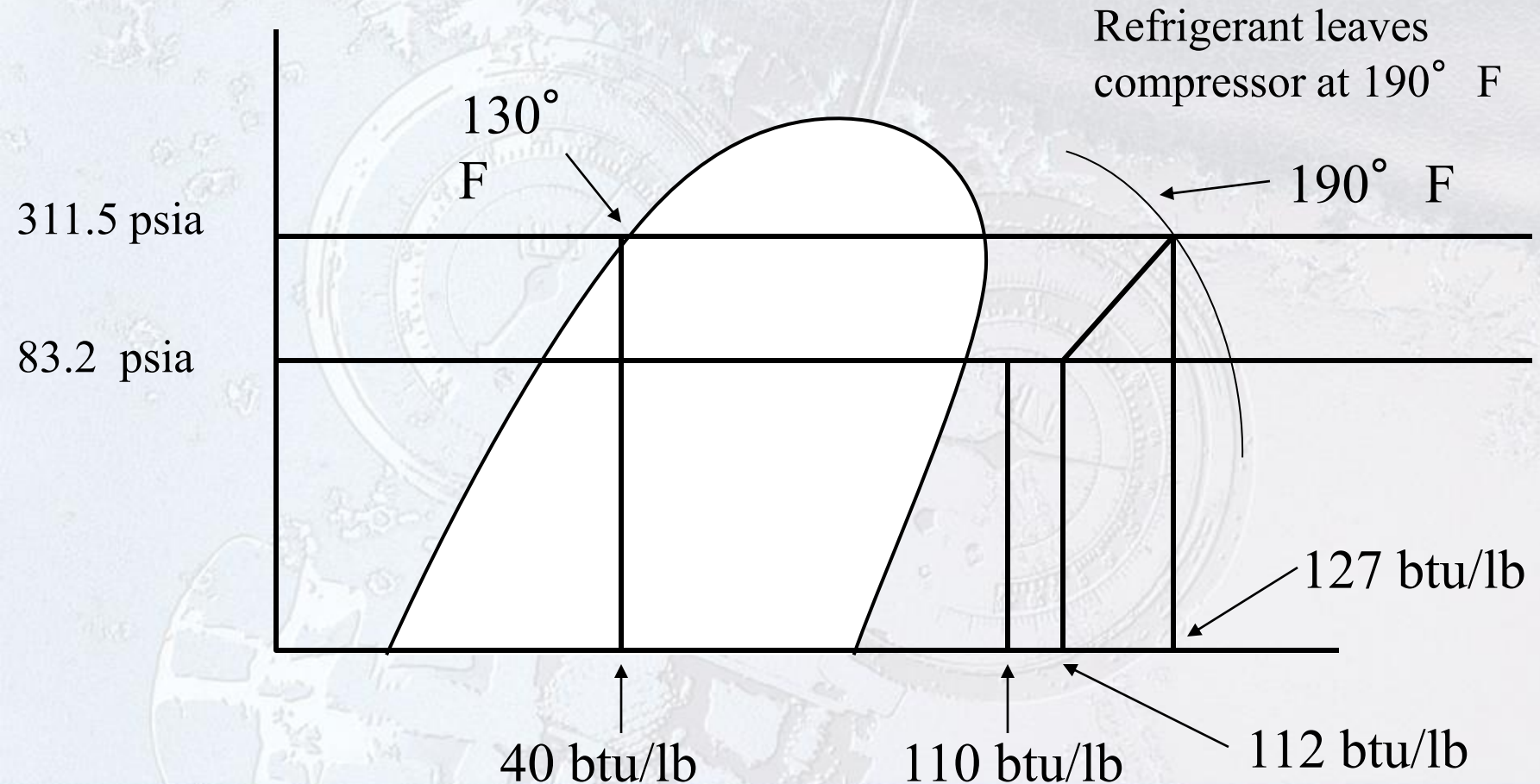
Suction line superheat is 10° F, so refrigerant enters compressor at 60° F



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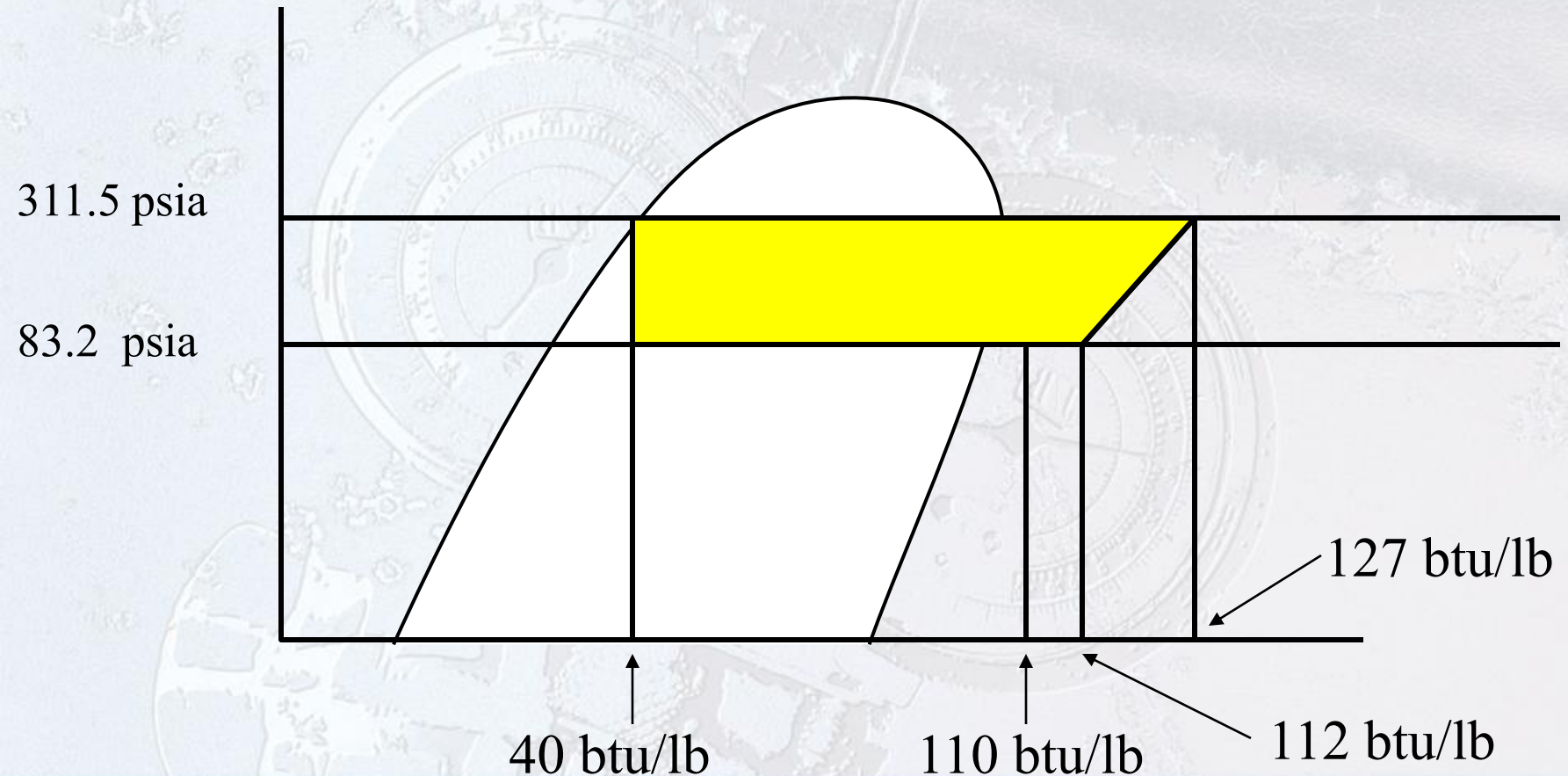
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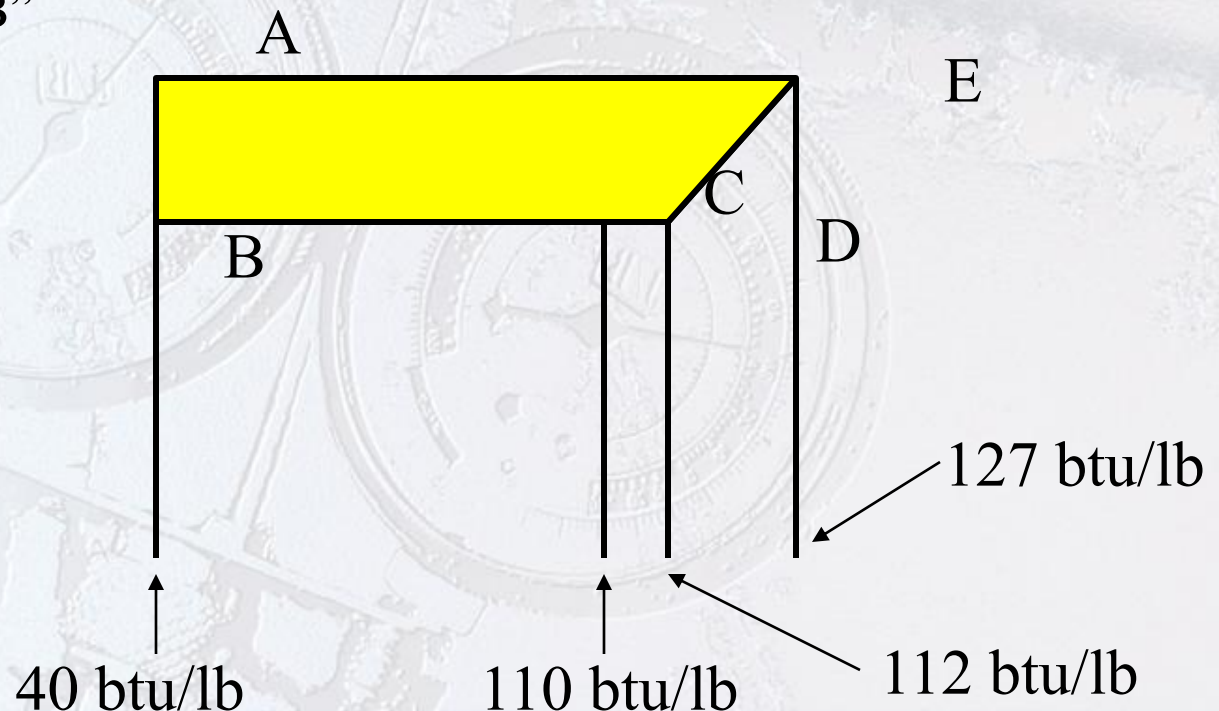
Metering device: “A to B”

Evaporator: “B to C”

Suction line: “C to D”

Compressor: “D to E”

Condenser “E to A”



PRESSURE ENTHALPY CALCULATIONS

- Net Refrigeration Effect (NRE) = $C - B = 110 \text{ btu/lb} - 40 \text{ btu/lb} = 70 \text{ btu/lb}$
- Heat of Compression (HOC) = $E - C = 127 \text{ btu/lb} - 110 \text{ btu/lb} = 17 \text{ btu/lb}$
- Total Heat of Rejection (THOR) = $E - A = 127 \text{ btu/lb} - 40 \text{ btu/lb} = 87 \text{ btu/lb}$
- Heat of Work = $E - D = 127 \text{ btu/lb} - 112 \text{ btu/lb} = 15 \text{ btu/lb}$