Instructor Contact Information

Instructor: Shuhsien Batamo, Ph.D.
Office: Central, Room LHSB 401
HCC Email: shuhsien.batamo@hccs.edu
Office Phone: 713-718-2544
Office Hours: By appointment only
Office Location: Learning Hub, Central Campus

Please note that in the College setting, there is no re-test or earning half of the credit backs by correcting the mistakes in the exams that were already taken if your score is not desirable. All students must earn their grade. If you miss the classes prior to the Official Date of Record, you’ll be automatically withdrawn from the class by HCC due to no show. Be aware of no-makeups policy per Departmental and Program rule.

All policies and class rules will be announced during the first day of class. So if you miss the first day of class, you miss quite a lot of information. If you’re dropped by the College, the instructor reserves the right NOT to reinstate you back into the class due to excessive absence.

Please feel free to contact me concerning any problems that you are experiencing in this course. Your performance in my class is very important to me. I am available to hear your concerns and just to discuss course topics.

HCC Email address required including preferred method of contact (e.g., email, phone); should you contact me, please use your HCC Student Email address http://50.56.172.79/district/students/student-e-mail-eagle-id/

Should you contact me by phone, please speak slowly concerning your message and phone number. I will respond to emails within 24 - 48 hours Monday through Friday; I will reply to weekend messages on Monday after 4 p.m. upon finishing up my classes.

What’s Exciting about This Course

You will learn so much about CHEMISTRY: The Central Science concerning the chemicals, medicines, energy, reactions and hands-on experiments to name a few that you encounter in your life. Please see Student Learning Outcome and Program Learning Outcome for details. The information in this course will enable you to understand why chemistry is the central science linking many disciplines and subjects of study.

My Personal Welcome

Welcome to General Chemistry I—I’m delighted that you have chosen this course. One of my passions is to know as much as I can about chemical reactions applicable to our daily life, and I can hardly wait to pass that on. I will present the information in the most exciting way I know, so that you can grasp the concepts and apply them now and hopefully throughout your life. As you read and wrestle with new ideas, theories, facts and applications that may challenge you, I am available to support you. The fastest way to reach me is by my HCC email on your syllabus. The best way to really discuss issues is in person and I’m available during break between classes and office hours by appointment, hopping around two campus locations, to tackle any questions you might have. My goal is for you to walk out of the course with a better understanding of chemistry and of chemical reactions and their applications in our life. So please visit me or contact me whenever you have a question.
Prerequisites and/or Co-Requisites

CHEM 1311 requires one-year high school chemistry and college-level reading and writing as well as Developmental Math 0312 or higher. Research indicates that you are most likely to succeed if you have already taken and passed the enrollment requirement courses: Reading 0342, Math 0312 and Writing 0310 /0349 or Math 0312 with INRW 0420 and taken prerequisites: One-year high school Chemistry or Introductory Chemistry lecture and lab. If you have enrolled in this course having satisfied these prerequisites, you have a higher chance of success than students who have not done so. Please carefully read and consider the repeater policy in the HCCS Student Handbook. http://www.hccs.edu/district/about-us/procedures/student-rights-policies-procedures/

Learning Web, Eagle Online Canvas, and Tutoring

This section of CHEM 1311 will use Learning Web (learning.hccs.edu) and Canvas to supplement in-class and/or online assignments, practice exams, quizzes and other activities. The usage of Mastering Chemistry is optional and are not monitored. Mastering Chemistry requires you to purchase the access code on top of textbook expense and thus is not mandatory.

Tutoring

HCC provides free, confidential, and convenient academic support to HCC students in an online environment and on campus. Tutoring is provided by HCC personnel in order to ensure that it is contextual and appropriate.

(1) In-person tutoring: visit http://hccs.edu/findatutor
(2) Online tutoring: visit http://hccs.upswing.io
(3) LibGuides for General & Organic Chemistry lecturing videos: visit http://library.hccs.edu/tutoring/chemorganic
(4) YouTube: Kham Academy & others (see instructor's postings in Canvas embedded in Power Point Slides).

Instructional Materials

o Textbook, Lab Manual, Reference, and Scientific Calculator, Scantron

(a) PDF version of eBook (from various online vendors including eBay, pricing ranging from $5-$20, contains 24 chapters for two semesters use CHEM 1311 and CHEM 1412 or hard copy are all acceptable. HCC custom version (containing 10-12 chapters for one semester use) is available only at HCC bookstore.
(b) Instructor does not use Mastering Chemistry. So used book, eBook, or older edition is all acceptable.

No smart watches and smart phones are allowed;

o Scantron: 882-E (4 required; Students need to purchase it from HCC bookstore.)
o Lab Manual: CHEM 1111 by Houston Community College Faculty (Blue Door). Available from HCC bookstore only.

The textbooks listed above are the images of the original textbooks from publisher 13th and 14th editions.

PDF version of 12th edition textbook and solutions manual are available for free download online by typing “free pdf download for Chemistry: The Central Science” by Brown/LeMay/Bursten et al. (Pearson).

o Nonprogrammable Scientific calculator without USB-port; TI-30 series ($8-$11) is the preferred one.

Other Resources: http://myeagle.hccs.edu to find the detailed information for the following list. Learning Web, Eagle Online Canvas, Find A Tutor, In Case of Emergency, Student Life, Student Help, Change Password, Library, Email, Class Search, Student Sign In (PeopleSoft), to name a few.
2.2 Starting with the reactants, complete the reaction by writing the reaction products.

2.1 Identify given reactions as combination, decomposition, single displacement, and double displacement.

SLO 2. Categorize, complete, and balance chemical reactions.

1.2 Given the name, write the formula of ionic compounds, binary molecular compounds, and acids. Given the formulas of these types of compounds, name them.

SLO 1. Give names and formulas of elements, ions, and ionic and molecular compounds.

1.1 Given the name, identify the formula and charge of positive and negative ions, and vice-versa.

2.1 Identify given reactions as combination, decomposition, single displacement, and double displacement.

2.2 Starting with the reactants, complete the reaction by writing the reaction products.
2.3 Given the reactants and products, balance the equation for the reaction.

SLO 3. Do chemistry calculations involving reaction stoichiometry and energy changes.
3.1 Convert amounts in units of mass or volume to moles, and vice-versa.
3.2 Given the amount of one substance in a reaction, calculate the amount of the other substances that react and form.
3.3 Identify the limiting reactant and excess reactant in a reaction where more than one reactant amount is given.
3.4 Determine the amount of the excess reactant that remains as unreacted excess.
3.5 Calculate energy changes associated with chemical reactions using Hess's law, standard enthalpies of formation, or calorimetry.

SLO 4. Relate the properties of electromagnetic radiation (frequency, wavelength, and energy) to each other and to the energy changes atoms undergo which accompany electronic transitions.
4.1 Relate frequency, wavelength, and the speed of electromagnetic radiation.
4.2 From the frequency or wavelength of electromagnetic radiation, calculate its energy.
4.3 Relate the energy change in the hydrogen atom to its electronic transitions using the Bohr model.
4.4 Identify and relate the four quantum numbers that can be associated with electrons.
4.5 Write the electronic configurations of atoms and ions, including the box diagram method.

SLO 5. Identify the parts of the periodic table and the trends in periodic properties of atoms. 5.1 Identify the common regions of the periodic table. Identify by name selected groups of elements in the periodic table.
5.2 Using the periodic table, identify the trend (increasing or decreasing in value) of selected properties of atoms such as atomic radius, ionization energy, and electron affinity.
5.3 Identify reaction similarities of elements within the same group in the periodic table.

SLO 6. Relate the properties of gases with the gas laws and extend the application of these relationships to reaction stoichiometry, gas mixtures, and effusion/diffusion of gases.
6.1 Relate and calculate the pressure, volume, temperature, or amount of gas using Boyle's law, Charles' law, Gay-Lussac's law, Avogadro's law, the combined gas law, and the ideal gas law.
6.2 Perform stoichiometry calculations which involve gaseous substances.
6.3 Use Dalton's law and Graham's law to perform calculations involving gaseous mixtures and effusion and diffusion of gases.
6.4 Explain the assumptions of the kinetic-molecular theory of gases.

SLO 7. Depict chemical bonding with dot structures and valence bond theory and determine the molecular shapes (geometry) of molecules based on VSEPR and valence bond theory.
7.1 Draw the Lewis dot structure of molecules containing two or more atoms.
7.2 Based on the dot structure of the molecule, determine its electron domain geometry and molecular geometry based on VSEPR theory.
7.3 Given the dot structure, identify the hybridization of and geometry about each atom.
7.4 Explain the nature of sigma and pi bonding using hybrid atomic orbitals.

SLO 8: Calculate density and relate the value to mass and volume measurements for all physical states.
8.1 Given either mass, volume, or density, be able to calculate an unknown variable through use of the density equation.
8.2 Appreciate the utility of density as an intensive and physical property as an identification tool.

SLO 9: Measurements and conversions in Metric, SI, and American systems
9.1 Convert and assess temperatures in three scales of measurement: Celsius, Fahrenheit, and Kelvin.
9.2 Convert measurements of mass, volume, length between established units of official International (SI), Metric, and American systems.

SLO 10: Apply thermochemical principles to evaluate work, heat, and energy relationships based on specific heat, calorimetry, and temperature changes.
10.1 Calculate heat based on mass, specific heat or heat capacity, and temperature change. 10.2 Understand the transfer of heat as it applies to a system and its surroundings, including calorimeters, by calculating one variable in an equation when presented with others including heat, mass, specific heat or heat capacity, and initial and final temperatures.
10.3 Define the meaning of work as it relates to energy in all forms: heat, potential and kinetic.
10.4 Apply the Law of Conservation of Energy as it pertains to energy exchange in thermochemical reactions. 10.5 Convert between SI and American units of heat.
Student Success in CHEM 1311

Chemistry is a math-based subject, which requires conceptual understanding and application, and is not a subject that you can learn or master passively. Chemistry is full of word problems and therefore mastering chemistry depends heavily on a person’s reading and math skills and of course the person’s determination.

Due to the content of this course, be prepared to spend considerable time in study. It is normal and common to spend around 20-25 hours or more per week studying for this course. The best practice is to spread the time evenly during the week instead of cramming it. Your dedication will reflect on how successful you’ll be in this class.

Chemistry is best learned through doing. Listening to lecture attentively and completion of homework assignment are essential for mastery of the course. Always come to class with preparation. Complete your reading assignment prior to the class and reviewing the class materials within 24 hours at home is the key for success.

CHEM 1311 is difficult and it’s a math-based and easy to fall behind. The adverse impacts on student performance include but not limit to frequent tardiness, early departure, absence and taking more than full-time student load (12 credit hours) while working full-time/overtime. This course is more rigorous than high school chemistry and Introductory Chemistry and thus requires students putting more effort than what they did in in their previous study.

The tests from the instructor are provided with reviews on Learning Web and are explained during the lecture and lab sections. Students are required to study the reviews and should students missed the review due to their absence, please ask for notes from other students. Scores from exams speak for student preparation for the class and how thorough the student has study for the tests.

Remember that reading solutions is solely different from solving problems and doing homework and practice exams yourself. To study effectively, you should read the assignments before attending lectures and lab.

Common Misconceptions among Students:

1. Students believe going to tutorial sections and watching the videos are equivalent to face-to-face instruction,
2. Tutors and videos are better and more knowledgeable than their professor,
3. The time spent with tutors and videos are more effective or equivalent to time spent in study and surely no idle time or time being wasted,
4. Coming to face-to-face lecture is boring and a waste of time, because the tutors (face-to-face or from the videos) know how to teach and my professor does not.
5. Texting and surfing internet is absolutely okay and is neither disruptive nor disrespectful to my fellow classmates and professor.
6. Cramming for chemistry before the test will guarantee a good grade.
7. Poor reading and math skills will not hinder my learning of chemistry as long as I am good in memorization.
8. Chemistry is not cumulative: if I am weak in nomenclature (chapter 2), stoichiometry (chapter 3), I can still do well in solution stoichiometry (chapter 4) and gas stoichiometry (chapter 10).

If you agree any one of the above, you’re strongly advised to drop the face-to-face class and enroll in the distance/online CHEM 1311 classes (16-wks, 12-wks, 1st 8-wks or 2nd 8-wks). There is more freedom in the online learning: students need not come to campus for the lecture for the entire semester. Students pick their time to watch the instruction delivered through videos and tests are done through online or in-person at the Testing Center.

Instructor and Student Responsibilities

As your Instructor, it is my responsibility to:

- Provide the grading scale and detailed grading formula explaining how student grades are to be derived
- Facilitate an effective learning environment through class activities, discussions, and lectures
- Provide a description of any special projects or assignments
- Inform students of policies such as attendance, withdrawal, tardiness and make up
- Provide the course outline and class calendar which will include a description of any special projects or assignments
- Arrange to meet with individual students before and after class as required
- Inform students instructor’s participation in Early Alert Program concerning student absence, academic readiness and performance

To be successful in this class, it is the student’s responsibility to:

- Attend class and participate in class discussions and activities
- Read and comprehend the textbook
• Complete the required assignments and exams:
• Ask for help when there is a question or problem
• Keep copies of all paperwork, including this syllabus, handouts, and all assignments
• Study the old system/departmental final exam reviews and exams provided by the Departmental Chair.
• Be aware of and comply with academic honesty policies in the HCCS Student Handbook http://www.hccs.edu/about-hcc/procedures/student-rights-policies--procedures/

Academic Integrity

You are expected to be familiar with the College’s Policy on Academic Honesty, found in the Student Handbook. What that means is: If you are charged with an offense, pleading ignorance of the rules will not help you. Students are responsible for conducting themselves with honor and integrity in fulfilling course requirements. Penalties and/or disciplinary proceedings may be initiated by College System officials against a student accused of scholastic dishonesty. “Scholastic dishonesty”: includes, but is not limited to, cheating on a test, plagiarism, and collusion. There is a Zero tolerance for any type of academic dishonesty. Please see the following link for further information: Student Handbook http://www.hccs.edu/resources-for/current-students/student-handbook/

Zero tolerance for academic dishonesty. Student who is caught cheating will receive a grade of zero for that exam or lab report with no exceptions and may be administratively withdrawn from the class. The student will be reported to the College for discipline action.

Plagiarism is academic dishonesty and is subjective to discipline action. Plagiarism includes the following conducts:
(a) taking the exam for the other student; taking pictures of any official exam(s) or Scantron(s)
(b) changing wrong answers to correct answers posted on Scantron,
(c) copying word-for-word for all assignment with same mistakes,
(d) photocopying or taking pictures of other student’s work and then wipe-outing the name, and claim as his/hers,
(e) turning in Scantron/exam with version written that does not match the version given and on student sign-in/attendance sheet,
(f) forging signatures by signing attendance sheet for another student(s), to name a few.

Student who is caught in plagiarism will receive a zero for that specific exam and be reported subjective to disciplinary action. The lecture exam grade zero as a result of plagiarism will neither be dropped nor replaced by System Final Exam.

Grading Policy

<table>
<thead>
<tr>
<th>Component</th>
<th>Weightage</th>
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</thead>
<tbody>
<tr>
<td>3 Lecture Exams (instructor’s exams)</td>
<td>75%</td>
</tr>
<tr>
<td>1 Comprehensive System Final Exam (Mandatory Departmental Exam)</td>
<td>25%</td>
</tr>
<tr>
<td>Total</td>
<td>100%</td>
</tr>
</tbody>
</table>

Final Average is rounded to a whole number according to method listed in chapter 1 to assign letter grade.
Grading Scale: 90-100 = A; 80-89= B; 70-79= C; 60-69= D; Below 60 = F; Students stop coming to class= FX. GPA calculation for FX is SAME as F. Please note: IP (in progress) does not apply to this course. For any incomplete grade, student must provide the document(s) and sign the Incomplete Grade Contract. Student who has no show for Departmental Final Exam will be issued a grade earned up to date.

Student letter grade is assigned to reflect how well the student has understood and fulfilled the requirement in the class, not what/how students feel they should get. Students who receive FX may face financial consequence by paying back the final aid money. Please visit http://www.hccs.edu/applying-and-paying/financial-aid/withdrawal-policy/ for details.

For withdrawal rules: http://ctle.hccs.edu/gcac/drop2.htm http://ctle.hccs.edu/gcac/faq2.htm Please note that withdrawal is solely student’s responsibility. Instructor does not drop students from the class for any reason. Please pay attention to the six course drop rule of the State of Texas (http://ctle.hccs.edu/gcac/faq2.htm) and the credit hours required by the INS office should you hold the F-1 visa or you are on financial aid status.

Exam Policy: All cell phones must be turned off and all smart watches must be removed from wrist. Leave both cell phones and smart watches in backpacks, and leave backpacks in front of the classroom. Should student carry concealed handgun, please keep it close to you in your pocket and be concealed.

No restroom use after 15 minutes of beginning of the exam unless the student has document on file with the Ability Office. No makeup exams are allowed for any excused and/or unexcused exam. Thus, students who missed the exam(s) are allowed to use their System Final Exam to replace their no-show lecture exam. Should students miss two lecture exams, they are advised to drop the class.
Do not take pictures of exams and Scantrons. *During or upon returning student Scantron with exam, if student copy or take the pictures of either exam(s) or Scantron(s), the student’s exam grade will be credited as Zero. All exams and Scantrons will be collected back and submitted to the chairman’s office at the end of semester.*

Should makeup exam is granted based on students’ excused slits, availability of class/lab room and non-instructional time, the students who took the makeup exam are not allowed to use the System Final Exam to replace their no-show exam.

Should the makeup exam is granted by providing instructor the excuse slit and taken within a week after the scheduled time, the maximum score is 80% of the exam grade earned; within two weeks, 70% of the exam grade earned; within three weeks, 60% of the exam grade earned; after three weeks, no makeup will be granted.

Please bring Scantron (FORM NO. 882-E), pencil, eraser, no-USB-port-equipped calculator and blank scratch paper to each test/exam. Write the version(s) and your name on your Scantron.

**CHEM 1311 Departmental Final Exam**

All students will be required to take a comprehensive departmental final exam consisting of 35 multiple-choice questions and 6 show-your-work questions. Students must provide their own Scantron forms (FORM NUMBER 882-E or 882-E-LOVAS). **Departmental/System Final Exam is NOT allowed to change and the schedule must be strictly followed according to the College Policy.** Should students have questions, please contact Department Chair.

**Policy Regarding Making-up Missed Assignments**

Per Department policy, no makeup labs and exams are allowed. Should it be granted due to an extreme condition, students must come to the instructor’s sections to do make-ups.

Please note that there is only one Chemistry classroom and one lab room at Stafford campus, which is used for all chemistry courses: Introductory Chemistry, General Chemistry I & II and Organic Chemistry I & II. There is only one General Chemistry lab room at Central which is used for Introductory Chemistry, General Chemistry I & II labs. Please understand that it’s difficult to match time and room on campus among three parties. Room-request requires application made to the Campus Manager in advance.

**HCC Holidays and Final Exam Schedule**

<table>
<thead>
<tr>
<th>Date</th>
<th>Event</th>
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<tbody>
<tr>
<td>1/14/2019</td>
<td>Reg. 16-Wk: Classes Begin</td>
</tr>
<tr>
<td>1/21/2019</td>
<td>Martin Luther King, Jr.</td>
</tr>
<tr>
<td>1/28/2019</td>
<td>16-Wk: Official Day of Record. Failure to show before this day will be automatically withdrawn from class by the College and will NOT be reinstated back to class. Please enroll in 2nd-start or 2nd 8-Wk class.</td>
</tr>
<tr>
<td>2/11/2019</td>
<td>2nd Start: Class Begin</td>
</tr>
<tr>
<td>2/18/2019</td>
<td>President’s day</td>
</tr>
<tr>
<td>3/11-3/17/2019</td>
<td>Spring Break</td>
</tr>
<tr>
<td>4/1/2019</td>
<td>16-Wk: Last Day for Student and Administrative Withdrawal</td>
</tr>
<tr>
<td>4/9/2019</td>
<td>12-Wk: Last Day for Student and Administrative Withdrawal</td>
</tr>
<tr>
<td>4/19-4/21/2019</td>
<td>Spring Holiday</td>
</tr>
<tr>
<td>5/6-5/11/2019</td>
<td>Final Exam Week: TBA (<a href="https://www.hccs.edu/student-experience/events-calendar/#/?i=1">https://www.hccs.edu/student-experience/events-calendar/#/?i=1</a>)</td>
</tr>
<tr>
<td>5/12/2019</td>
<td>Both 2nd Start and 16-Wk: Semester Ends</td>
</tr>
</tbody>
</table>

**Course Calendar: Assignments shown here are not graded and are based on 13th edition.**

<table>
<thead>
<tr>
<th>Date</th>
<th>Day</th>
<th>Topic</th>
</tr>
</thead>
<tbody>
<tr>
<td>1/15</td>
<td>T</td>
<td>Syllabus &amp; Announcement</td>
</tr>
<tr>
<td>1/17</td>
<td></td>
<td>Chap 1: starting p. 33: 1.1, 1.2, 1.4, 1.6, 1.8, 1.13, 1.15(b)(c)(d), 1.16(e)(f)(g)(i), 1.21, 1.23, 1.25, 1.27, 1.31(b)(d), 1.37, 1.39, 1.41(a)(d), 1.42(a), 1.45(a)(b)(d)</td>
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<tr>
<td>1/22</td>
<td></td>
<td>Chap 1 (finished)</td>
</tr>
<tr>
<td>1/24</td>
<td></td>
<td>Chap 2: starting p. 73: 2.3, 2.4, 2.5, 2.6, 2.7, 2.9, 2.23, 2.26, 2.27(a)(b), 2.28(c)(e), 2.29, 2.35, 2.39(a), 2.41, 2.45, 2.47, 2.51, 2.53, 2.55, 2.57, 2.59, 2.61, 2.63, 2.65, 2.69, 2.71, 2.73, 2.75(b)(c), 2.76(a)(c)(d)(e)(f), 2.77, 2.82, 2.83, 2.86. Difficult section: Naming/chemical formula, relying on memorization.</td>
</tr>
<tr>
<td>1/28</td>
<td>M</td>
<td>Official Date of Record.</td>
</tr>
<tr>
<td>1/29</td>
<td>T</td>
<td>Chap 2 (finished)</td>
</tr>
<tr>
<td>Date</td>
<td>Day</td>
<td>Activity</td>
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<tr>
<td>2/5</td>
<td>T</td>
<td>Chap 3 (continued)</td>
</tr>
<tr>
<td>2/7</td>
<td></td>
<td>Chap 3 (finished)</td>
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<tr>
<td>2/14</td>
<td></td>
<td>Exam I (Chapters 1 to 3) 70 minutes only. Bring Calculator, Scantron, pencil, eraser.</td>
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<tr>
<td>2/19</td>
<td>T</td>
<td>Chap 4 (continued)</td>
</tr>
<tr>
<td>2/21</td>
<td></td>
<td>Chap 4 (continued)</td>
</tr>
<tr>
<td>2/26</td>
<td>T</td>
<td>Chap 5: starting p. 202: 5.4, 5.19, 5.21, 5.23, 5.25, 5.39(a), 5.43, 5.45, 5.49, 5.51(a)(b)(d), 5.53(a), 5.57, 5.59(a), 5.63, 5.65, 5.66, 5.68(b), 5.73, 5.75</td>
</tr>
<tr>
<td>2/28</td>
<td></td>
<td>Chap 5 (finished)</td>
</tr>
<tr>
<td>3/5</td>
<td>T</td>
<td>Chap 6: starting p. 248: 6.7(c), 6.11, 6.12, 6.25, 6.37(a), 6.55, 6.57, 6.59, 6.61, 6.62, 6.63, 6.64, 6.70, 6.71, 6.72, 6.73, 6.75(a)(b), 6.76(a)(c)(d), 6.77, 6.78(a)(b), 6.79, 6.80, 6.93</td>
</tr>
<tr>
<td>3/19</td>
<td>T</td>
<td>Chap 7: starting p. 289: 7.2, 7.3, 7.4, 7.7, 7.8, 7.13, 7.17, 7.25, 7.27, 7.28, 7.29, 7.31(a), 7.34, 7.35, 7.39(a), 7.41, 7.43, 7.45, 7.55(a)(b), 7.59, 7.61. Key point for Chapter 7: sect. 7.2-7.5; others are repetitive from chapter 2.</td>
</tr>
<tr>
<td>3/21</td>
<td></td>
<td>Exam II (chapters 4 to chap 6).</td>
</tr>
<tr>
<td>3/26</td>
<td>T</td>
<td>Chap 8: starting p. 333: 8.1, 8.2, 8.4, 8.7, 8.9, 8.11, 8.14(a)(b)(d), 8.17, 8.198.21(a), 8.23 (memorize the lattice energy trend for these 3 compounds); 8.25(a), 8.26, 8.35, 8.37, 8.38, 8.39, 8.41, 8.47, 8.48(a)(b)(c), 8.52, 8.53(a)(c), 8.55, 8.57, 8.59, 8.60; **8.63 [(e) is missing 2-], 8.64(b)(c)[e]65, 8.69, 8.71.</td>
</tr>
<tr>
<td>4/1</td>
<td>M</td>
<td>Last Day for Student &amp; Administrative Withdrawal.</td>
</tr>
<tr>
<td>4/28</td>
<td></td>
<td>Chap 8 (continued)</td>
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<tr>
<td>4/9</td>
<td>T</td>
<td>Chap 8 (finished)</td>
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<tr>
<td>4/11</td>
<td></td>
<td>Chap 9 (continued)</td>
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<tr>
<td>4/16</td>
<td>T</td>
<td>Chap 9 (finished)</td>
</tr>
<tr>
<td>4/18</td>
<td>T</td>
<td>Chap 10: starting p. 432: 10.19(a)(b), 10.20(a)(c), 10.23, 10.29, 10.26, 10.33, 10.34(a)(b)(c), 10.37, 10.41, 10.43(a), 10.47, 10.49, 10.51, 10.55, 10.57, 10.59, 10.63, 10.73, 10.75, 10.81, 10.89, 10.91</td>
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<tr>
<td>4/23</td>
<td>T</td>
<td>Chap 10 (finished)</td>
</tr>
<tr>
<td>4/25</td>
<td></td>
<td>Exam III (chap 7, 8, 9)</td>
</tr>
<tr>
<td>5/2</td>
<td></td>
<td>Review for System Final Exam (bring questions, no re-lecture).</td>
</tr>
<tr>
<td>5/7</td>
<td>T</td>
<td>Please refer to the updated Final Exam information on <a href="https://www.hccs.edu/student-experience/events-calendar/#/?i=1">https://www.hccs.edu/student-experience/events-calendar/#/?i=1</a> System/Departmental Comprehensive Final Exam (11:00 a.m. – 1:00 p.m.)</td>
</tr>
<tr>
<td>5/13</td>
<td>M</td>
<td>Grade Available to Students</td>
</tr>
</tbody>
</table>

**Syllabus Modifications**

The instructor reserves the right to modify the syllabus at any time during the semester and will promptly notify students in writing, typically by e-mail, of any such changes.
The EGLS$^3$ (Evaluation for Greater Learning Student Survey System) will be available for most courses near the end of the term until finals start. This brief survey will give invaluable information to your faculty about their teaching. Results are anonymous and will be available to faculty and division chairs after the end of the term. EGLS$^3$ surveys are only available for the Fall and Spring semesters. EGLS surveys are not offered during the Summer semester due to logistical constraints.

http://www.hccs.edu/resources-for/current-students/egls3-evaluate-your-professors/

HCC Email Policy
HCC prefers students to communicate only through the HCCS email system to protect your privacy. If you have not activated your HCCS student email account, you can go to HCC Eagle ID and activate it now. Instructor uses HCC email to communicate.

HCC Policy Statements
Here's the link to the HCC Student Handbook http://www.hccs.edu/resources-for/current-students/student-handbook/ Though it's labeled 2015-2016 Student Handbook, it's basically applicable to current academic year. In it you will find information about the following:

Academic Honesty
Academic Information
Academic Support
Attendance, Repeating Courses, and Withdrawal
Campus Carry
Career Planning and Job Search
Childcare
Course Etiquette
disAbility Support Services
Electronic Devices
Equal Educational Opportunity
Financial Aid TV (FATV)
General Student Complaints
Grade of FX and International Students
Health Awareness
Incomplete Grades
International Student Services
Libraries/Bookstore
Police Services & Campus Safety
Student Life at HCC
Student Rights and Responsibilities
Student Services
Testing
Transfer Planning
Veteran Services

Other Announcements from HCCS:

(1) Active Shooter Event (Ready Houston) https://www.youtube.com/watch?v=5VcSwejU2D0
Active Shooter in School https://www.youtube.com/watch?v=VCd6SAeWbHM

(2) FERPA (Family Educational Rights and Privacy Act): Should a student under such protection, student must initiate the initial contact with the instructor in person by showing photo ID and verifying phone number and email address. Only HCC email address is allowed to use for communication.

(3) Campus Carry in Texas: “At HCC the safety of our students, staff, and faculty is our first priority. As of August 1, 2017, Houston Community College is subject to the Campus Carry Law (SB11 2015). For more information, visit the HCC Campus Carry web page at http://www.hccs.edu/departments/police/campus-carry/ Please pay attention to the sign posted on the building or room concerning the campus carry law.

Please know the difference between the campus carry and open carry. It is illegal to openly carry on campus.
http://www.hccs.edu/departments/police/campus-carry/campus-carry-and-open-carry-faqs/ HCC Police Department Phone Number: 713-718-8888. Should you carry a concealed handgun to campus, please keep it near/with your body and keep it concealed at all times. This will prevent you violate the law.

(4) Title IX of the Education Amendments of 1972 ("Title IX"), http://www.hccs.edu/departments/institutional-equity/title-ix-know-your-rights/ “Houston Community College is committed to cultivating an environment free from
inappropriate conduct of a sexual or gender-based nature including sex discrimination, sexual assault, sexual harassment, and sexual violence. Sex discrimination includes all forms of sexual and gender-based misconduct and violates an individual’s fundamental rights and personal dignity. Title IX prohibits discrimination on the basis of sex-including pregnancy and parental status-in educational programs and activities. If you require an accommodation due to pregnancy please contact an Abilities Services Counselor. The Director of EEO/Compliance is designated as the Title IX Coordinator and Section 504 Coordinator. All inquiries concerning HCC policies, compliance with applicable laws, statutes, and regulations (such as Title VI, Title IX, and Section 504), and complaints may be directed to:

David Cross, Director EEO/Compliance Office of Institutional Equity & Diversity 3100 Main (713) 718-8271 Houston, TX 77266-7517 or Houston, TX 77266-7517 or http://www.hccs.edu/departments/institutional-equity/"

(6) American Disability Act: http://www.hccs.edu/support-services/disability-services/
HCC strives to make all learning experiences as accessible as possible. If you anticipate or experience academic barriers based on your disability (including mental health, chronic or temporary medical conditions), please meet with a campus Abilities Counselor as soon as possible in order to establish reasonable accommodations. Reasonable accommodations are established through an interactive process between you, your instructor(s) and Ability Services. It is the policy and practice of HCC to create inclusive and accessible learning environments consistent with federal and state law.

The Office of Students with at HCC reminds faculty that they are required to include the following statement on all their class syllabi: Any student with a documented disability (e.g. physical, learning, psychiatric, vision, hearing, etc.) who needs to arrange reasonable accommodations must contact the Ability Office at the respective college at the beginning of each semester. Faculty is authorized to provide only the accommodations requested by the Ability Office. The accommodation is not retroactive and thus student who needs this accommodation, please contact the Ability Office in a timely manner.

Meet the ADA Counselors and Support Staff:
http://www.hccs.edu/support-services/disability-services/ada-counselors/

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