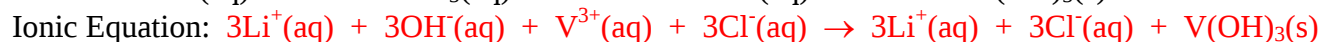
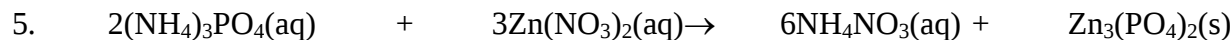
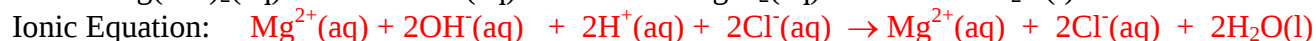
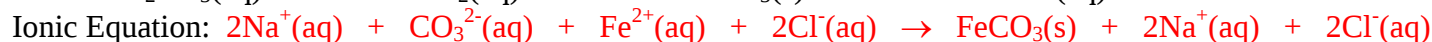
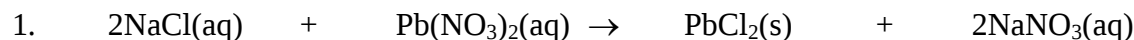
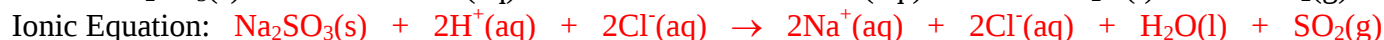
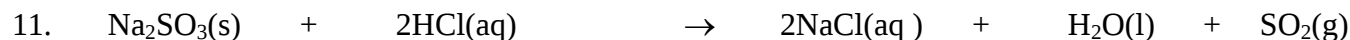
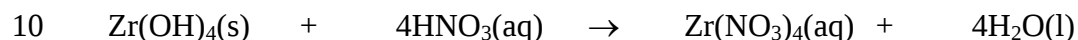
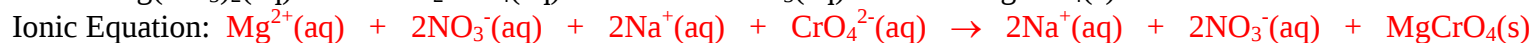
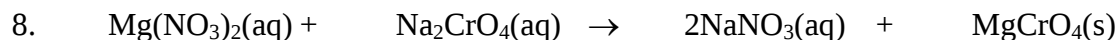
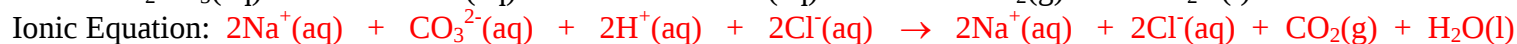
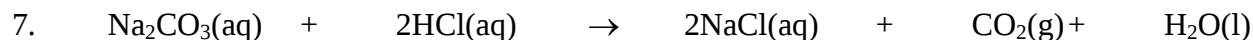


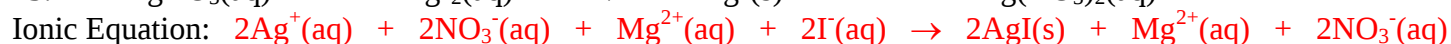
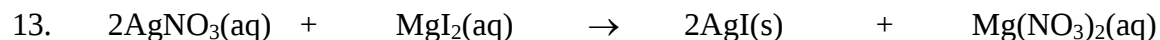
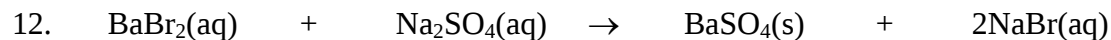
Net Ionic Equation Worksheet Answers

Write balanced molecular, ionic, and net ionic equations (NIE) for each of the following reactions. Assume all reactions occur in aqueous solution.





(note that the sodium ions do not cancel out since they are part of the solid phase in the reactants and part of the aqueous phase in the products)



(your final answer would be: $\text{Ag}^+(\text{aq}) + \text{I}^-(\text{aq}) \rightarrow \text{AgI}(\text{s})$)

